

# An Aromatic Universe - Low Temperature Molecular Mass Growth Processes to Polycyclic Aromatic Hydrocarbons (PAHs)

Ralf I. Kaiser<sup>1,2</sup>

<sup>1</sup>*Department of Chemistry, University of Hawaii at Manoa, Honolulu HI 96822, USA*

<sup>2</sup>*W.M. Keck Research Laboratory in Astrochemistry, University of Hawaii at Manoa, Honolulu HI 96822, USA*

Polycyclic aromatic hydrocarbons (PAHs) represent key molecular building blocks leading to carbonaceous nanoparticles identified in combustion systems and extraterrestrial environments. However, the understanding of their formation and growth has remained elusive. Here, we present evidence based on laboratory data combined with electronic structure calculations on fundamental mass growth processes of PAHs in the gas phase via ring expansion and ring annulation. Key reaction pathways operate at ultralow temperatures such as at 10 K as present in cold molecular clods like TMC-1 synthesizing PAHs at least from two to five six membered rings such as naphthalene, anthracene, phenanthrene, triphenylene, [4]-helicene, and [5]-helicene. These elementary reactions are rapid, have no entrance barriers, and synthesize PAHs via van-der-Waals complexes and submerged barriers. This facile route to complex PAHs signifies a critical shift in the perception that PAHs can be only formed at high-temperature combustion and circumstellar conditions providing a detailed understanding of the low temperature chemistry through untangling elementary reactions on the most fundamental level. An outlook is also presented on the synthesis of PAHs in low temperature, hydrocarbon-dominated ices in deep space and in the outer Solar System upon interaction with (non)-ionizing radiation; these processes are driven by non-adiabatic reaction dynamics and low lying triplet states of acetylene leading to PAHs as complex as coronene. Overall, these mechanisms eventually lead to graphene-type PAHs and two dimensional nanostructures providing an exciting view about the transformation of carbon in our universe.

## References

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