

Carbon Isotope and Isotopomer Fractionation in Cold Dense Cloud Cores

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¹³C should be useful to investigate chemistry of carbon bearing species. Recent observations in TMC-1 indicated that the molecular abundances of carbon isotopomers are different. Takano et al. (1998) observed HC₃N and found HCC¹³CN is more abundant than HC¹³CCN and H¹³CCCN, which indicates three carbon atoms are not equivalent in HC₃N. Sakai et al. (2007; 2010) reported the abundance ratios of C¹³CS/¹³CCS = 4.2 and CCH/¹³CCH = 1.6. Again, these results indicate two carbon atoms are not equivalent in CCS and CCH. They pointed out there are two possible processes to cause these fractionation: (i) the formation path ways of the species and (ii) the exchange of the ¹³C position after formation of molecules by isotopomer-exchange reactions.

We construct the gas-grain chemical network model which includes carbon isotopes (¹²C and ¹³C). Temporal variations of molecular abundances, the carbon isotope ratios (¹²CX/¹³CX) and the isotopomer ratios (¹²C¹³CX/¹³C¹²CX) of CCH and CCS in cold dense cloud cores are investigated by numerical calculations.

We reproduce the observed C¹³CH/¹³CCH ratio in TMC-1; isotopomer-exchange reaction, ¹³CCH + H → C¹³CH + H. However, the C¹³CS/¹³CCS ratio is lower than observed in TMC-1. We propose the isotopomer-exchange reaction, ¹³CCS + H → C¹³CS + H + 15K. In the model with this reaction, we reproduce the observed C¹³CS/¹³CCS, CCS/C¹³CS and CCS/¹³CCS ratio simultaneously.

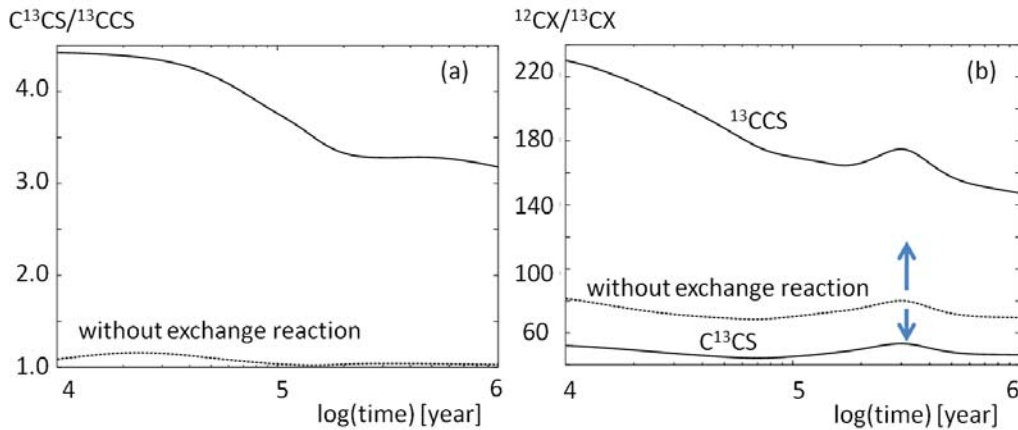


Figure 1: Temporal variation of (a) isotopomer ratios: the C¹³CS/¹³CCS ratio and (b) isotope ratios of CCS: the CCS/¹³CCS and CCS/C¹³CS ratios. Solid lines show ratios in the model with the reaction, ¹³CCS + H → C¹³CS + H + 15K. Dashed line shows the isotope ratio of CCS in the model without the reaction. The density is n_H = 10⁵ cm⁻³.

References

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