## Vibrational Circular Dichroism of a Chiral Triplet Nitrene Investigated under Matrix-Isolation Conditions in para-Hydrogen

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Vibrational circular dichroism (VCD) spectroscopy is an established method for the determination of absolute configurations (AC). <sup>1</sup> It measures the difference in absorbance between left- and right-circularly polarized light by vibrational transitions. The signal intensity of a transition is given by the rotational strength which depends on both the electronic (ETDM) and the magnetic (MTDM) transition dipole moment.

Therefore, VCD spectra of chiral high-spin organic molecules are expected to show a strong intensity enhancement due to the influence of unpaired electrons on the MTDM. <sup>2</sup> These influences are not assumed to be correctly predicted by the magnetic field perturbation (MFP) theory which was originally developed for closed-shell systems and is implemented in current computational protocols. <sup>3</sup>

In our study, we isolated chiral triplet nitrene precursor 2-azido-9*H*-fluorenol **1** in parahydrogen matrices and investigated its properties by matrix-isolation infra-red/vibrational circular dichroism (MI-IR/VCD) spectroscopy. MI-VCD spectra of **1** show satisfactory mirrorimage quality and are in agreement with density functional theory (DFT) calculations at B3LYP/6-311++G(2d,p) level of theory. Irradiation at 254 nm formed the triplet nitrene 2-nitreno-9*H*-fluorenol **T-2** whose spectral features are reproduced unexpectedly well by DFT calculations and do not exhibit the anticipated signal intensity enhancement.

Thereby, our experiments demonstrated the general feasibility of MI-IR/VCD spectroscopic studies on chiral high-spin organic molecules and the computational reproducibility of their spectral features.



Figure 1: Structure of 2-azido-9*H*-fluorenol 1 and the triplet nitrene T-2 resulting from photochemical cleavage of  $N_2$ .

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