

Theoretical exploration of astrochemical reactions: Leveraging ring-polymer molecular dynamics and machine-learning techniques

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We have embarked on the research project entitled “Next Generation Astrochemistry”. Our mission is to comprehend reaction mechanisms¹⁻⁷ and furnish reliable thermal rate coefficients to enhance the precision of the rate coefficients in the astrochemical database.

To understand chemical reactions, a potential energy surface spanning the $3N-6$ dimensions, where N represents the number of atoms, is necessary. We have recently developed several full-dimensional potential energy surfaces (PESs) such as $\text{H} + \text{C}_2\text{H}_2 \rightarrow \text{H}_2 + \text{C}_2\text{H}^\cdot$, $\text{NH}_3^+ \text{H}_2 \rightarrow \text{NH}_4^+ + \text{H}$, and $(\text{HCOOH})\text{H}^+ \rightarrow \text{CO} + \text{H}_3\text{O}^+ / \text{HCO}^+ + \text{H}_2\text{O}$ reactions¹⁻³. To develop these PESs, we employed a machine-learning technique, utilizing polynomial fitting from preliminary dynamics data and potential energies at various intrinsic reaction coordinates.⁴

In the cold environments typical of the interstellar medium, nuclear quantum effects can profoundly impact chemical dynamics. Hence, we employed the ring-polymer molecular dynamics (RPMD) simulation method. Our investigations have revealed that nuclear quantum effects and the quantum fluctuation of nuclei play a crucial role not only in the aforementioned collision reactions¹⁻³ but also in determining the H_2 sticking probability on ice clusters.⁵

In this presentation, we will introduce our recent research on the branching dynamics of $\text{H}_3^+ + \text{C}_2\text{H}_4 \rightarrow \text{H}_2 + \text{C}_2\text{H}_5^+ / 2\text{H}_2 + \text{C}_2\text{H}_3^+$ and $\text{H}_3^+ + \text{HNCO} \rightarrow \text{H}_2 + \text{HNCOH}^+ / \text{H}_2 + \text{H}_2\text{NCO}^+$ reactions. Additionally, we delve into the transition state dynamics of $\text{CH}_3\text{OH} + \text{OH}^\cdot \rightarrow \text{CH}_3\text{O}^\cdot + \text{H}_2\text{O}$ with the goal of comprehending the intricate reaction mechanism post-transition state from a dynamical viewpoint, a perspective unattainable through the statistical transition state theory.

¹ Murakami, T.; Iida, R.; Hashimoto, Y.; Takahashi, Y.; Takahashi, S.; Takayanagi, T. Ring-Polymer Molecular Dynamics and Kinetics for the $\text{H} + \text{C}_2\text{H}_2 \rightarrow \text{H}_2 + \text{C}_2\text{H}^\cdot$ Reaction Using the Full-Dimensional Potential Energy Surface. *J. Phys. Chem. A* **2022**, *126*, 9244-9258.

² Hashimoto, Y.; Takayanagi, T.; Murakami, T. Theoretical Calculations of the Thermal Rate Coefficients for the Interstellar $\text{NH}_3^+ \text{H}_2 \rightarrow \text{NH}_4^+ + \text{H}$ Reaction on a New Δ -Machine Learning Potential Energy Surface. *ACS Earth Space Chem.* **2023**, *7*, 623-631.

³ Murakami, T.; Ibuki, S.; Hashimoto, Y.; Kikuma, Y.; Takayanagi, T. Dynamics study of the post-transition-state-bifurcation process of the $(\text{HCOOH})\text{H}^+ \rightarrow \text{CO} + \text{H}_3\text{O}^+ / \text{HCO}^+ + \text{H}_2\text{O}$ dissociation: Application of machine-learning techniques. *Phys. Chem. Chem. Phys.* **2023**, *25*, 14016-14027.

⁴ Takayanagi, T. Application of Reaction Path Search Calculations to Potential Energy Surface Fits. *J. Phys. Chem. A* **2021**, *125*, 3994-4002.

⁵ Murakami, T.; Ogino, K.; Hashimoto, Y.; Takayanagi, T. Ring-polymer Molecular Dynamics Simulation for the Adsorption of H_2 on Ice Clusters $(\text{H}_2\text{O})_n$ ($n=8, 10, \text{ and } 12$). *ChemPhysChem.* **2023**, *24*, e202200939.

⁶ Murakami, T.; Takayanagi, T. Interstellar Benzene Formation Mechanisms via Acetylene Cyclotrimerization Catalyzed by Fe^+ Attached to Water Ice Clusters: Quantum Chemical Calculation Study. *Molecules.* **2022**, *27*, 7767.

⁷ Murakami, T.; Matsumoto, N.; Fujihara, T.; Takayanagi, T. Possible Roles of Transition Metal Cations in the Formation of Interstellar Benzene via Catalytic Acetylene Cyclotrimerization. *Molecules.* **2023**, *28*, 7454.