## Electronic Spectroscopy of Protonated and Hydrogenated Coronene in Solid *Para*-Hydrogen

Weber, I.,<sup>1</sup>\* Lee, Y.-P.<sup>2</sup>

<sup>1</sup> iweber@nycu.edu.tw, Department of Applied Chemistry, National Yang-Ming Chiao-Tung University, Hsinchu 300093, Taiwan.

<sup>2</sup> Department of Applied Chemistry, National Yang-Ming Chiao-Tung University, Hsinchu 300093, Taiwan and Center for Emergent Functional Matter Science, National Yang-Ming Chiao-Tung University, Hsinchu 300093, Taiwan.

Although well over 500 diffuse interstellar bands (DIB) have been identified up to today, their carrier molecules remain mostly elusive: only the buckminsterfullerene cation  $C_{60}^+$  has been confirmed as the carrier of five DIB in the near IR.<sup>1</sup> The most promising candidates considered include polycyclic aromatic hydrocarbons (PAH) and their cationic, protonated and hydrogenated derivatives; however, low-temperature gas-phase electronic absorption spectra of these molecules are rarely available.

Solid *para*-hydrogen (*para*-H<sub>2</sub>) has frequently been employed as a matrix host to record IR absorption spectra of protonated and hydrogenated PAH, which can be conveniently prepared by H<sup>+</sup>-transfer reaction with H<sub>3</sub><sup>+</sup> generated by electron bombardment during deposition. As a quantum solid, *para*-H<sub>2</sub> is soft and can thus provide a homogenous environment leading to only small matrix shifts. Over the past years, we have studied the dispersed fluorescence and fluorescence excitation spectra of several neutral, protonated, and hydrogenated PAH and PANH ( $N_{\rm C} < 42$ ) and found consistent but comparably small red shifts (40–110 cm<sup>-1</sup>) with respect to literature gas-phase spectra.

Here, we present the dispersed fluorescence and fluorescence excitation spectra of coronene (C<sub>24</sub>H<sub>12</sub>) and its most stable protonated (1-H<sup>+</sup>C<sub>24</sub>H<sub>12</sub>) and hydrogenated (1-HC<sub>24</sub>H<sub>12</sub>) isomers. Upon deposition of C<sub>24</sub>H<sub>12</sub> and *para*-H<sub>2</sub> under electron bombardment, we observed two distinct emission systems with strong origin bands at ~14335 and ~12820 cm<sup>-1</sup>, which we assign to 1-H<sup>+</sup>C<sub>24</sub>H<sub>12</sub> and 1-HC<sub>24</sub>H<sub>12</sub>, respectively, based on lowest-energy transition energies predicted by TD-DFT calculations (716.51 nm for 1-H<sup>+</sup>C<sub>24</sub>H<sub>12</sub> and 753.45 nm for 1-HC<sub>24</sub>H<sub>12</sub>) and the reasonable agreement of the experimental spectra with simulated emission spectra. For the  $S_1$ -S<sub>0</sub> transition of 1-H<sup>+</sup>C<sub>24</sub>H<sub>12</sub>, our data indicates a red shift of ~40 cm<sup>-1</sup> relative to the gas-phase<sup>2</sup> due to the *para*-H<sub>2</sub> environment, consistent with our results for other PAH. Protonation and hydrogenation lead to a significant redshift compared to the parent C<sub>24</sub>H<sub>12</sub> (S<sub>1</sub>-S<sub>0</sub> at ~23720 cm<sup>-1</sup> in *para*-H<sub>2</sub>) and a shortening in radiative lifetimes. Detailed assignments for the dispersed fluorescence and fluorescence excitation spectra are derived, illustrating the impact of symmetry breaking due to hydrogenation and protonation, and a comparison to available literature data from cryogenic ion-trap<sup>2</sup> and Ne matrix isolation<sup>3,4</sup> experiments is provided. The dispersed fluorescence spectrum of 1-HC<sub>24</sub>H<sub>12</sub> has not been reported previously.

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